



Project status

Simon Markthaler



Funded by the European Union HORIZON EUROPE grant agreement Nº 101084066 Institute of Energy Process Engineering Friedrich-Alexander-Universität Erlangen-Nürnberg



Department of Chemical and Bioengineering (CBI) • Chair of Energy Process Engineering • Prof. Dr.-Ing. Jürgen Karl • Prof. Dr. Katharina Herkendell

CarbonNeutralLNG



Status Quo

From renewable electricity!

hydrogen

Folie 2

- Energy density of batteries is still limited
- The worldwide conversion of the existing infrastructure will certainly take decades
- **GreenLNG** is certainly the most adequate option for the marine sector

carbon

From biomass!



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"e-fuels"





The CarbonNeutralLNG Process



Folie 3







The CarbonNeutralLNG Process



Folie 5



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Advanced Process Control









Legal, Environmental, and Economical approach:

- Impact assessment at environmental and socio-economic level
- Roadmap to replication: market and policy assessment
- Coordination and Dissemination





Main Role of the TTDM Board





Erlangen-Nürnberg

Development of an **additively manufactured** catalytic methanation reactor with **in-situ tar co-reforming**

Jakob Müller, Simon Markthaler Institute of Energy Process Engineering Friedrich-Alexander-Universität Erlangen-Nürnberg







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Outline

- 1. Methanation and co-reforming of tars
- 2. Preliminary designs of 3D-printed methanation reactors
- **3.** Novel methanation reactor designs
- 4. Outlook







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Tar development in biomass gasification

• **Tars** are **aromatic compounds** that are formed during the thermochemical gasification of biomass



[1] Font Palma, C. (2013). Modelling of tar formation and evolution for biomass gasification: A review. In Applied Energy (Vol. 111, pp. 129–141). Elsevier Ltd. https://doi.org/10.1016/j.apenergy.2013.04.082





Problem with tar contamination

- Tars condense or crystallize at low temperatures, which can cause problems in plant engineering (deposits, blockages, etc.).
- Tars are **problematic for archaea** in biological methanation.







(Steam reforming)

(Dry reforming)

Removal of tars:

- Classical gas cleaning is expensive and complex
- Alternative: steam reforming, dry reforming, or thermal cracking
 - For instance, Benzene: $C_6H_6 + 6H_2O \rightarrow 6CO + 9H_2$

 $C_6H_6 + 6\ CO_2 \rightarrow 12\ CO + 3\ H_2$

• The reactions are **endothermic** and occur **at high temperatures**



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Folie 18

Co-reforming of tars in a methanation reactor (Advanced geometry by 3D printing to tailor temperatures**)**

+ Excess steam/hydrogen





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Temperature profile of methanation Methanisierungsreaktion

- At the **inlet** of the reactor ①, **higher temperatures** are advantageous to improve the kinetics. The maximum operating temperature of the catalyst must be considered.
- In the **high-temperature** zone (2), there are good conditions for the **reforming of tars**



At the outlet of the reactor 3, the temperatures should drop to around 250°C to achieve a high methane yield.



ADDmeth1 via additive manufacturing

- The **reaction channel** is **widened** at the top of the reactor to prevent kinetic limitation at the outlet
- The conical reaction channel maintains temperatures at the outlet at the required level.



• Heatpipes for cooling

ADDmeth1 via additive manufacturing



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ADDmeth1: Results on synthesis power

- The reactor exhibits the desired temperature profile
- However, at higher power levels, the hotspot shifts towards the rear
- The hotspot in the lower section does not yet have the desired profile









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ADDmeth1: Results on synthesis power

- The reactor exhibits the desired temperature profile (during pure CO2 methanation)
- However, at higher power levels, the hotspot shifts towards the rear
- The hotspot in the lower section does not yet have the desired profile







Variant 1 (ADDmeth3.1)

- The heat pipe is **centrally positioned**, and the reaction channel is arranged concentrically around it
- The cross-section in the lower region is larger, leading to an increase in local residence time
- Using a single heat pipe instead of three enhances the heat dissipation design







Variant 2 (ADDmeth3.2)

- Three heat pipes with reduced diameter are arranged in a cloverleaf structure
- The heat pipes follow the widening of the reaction channel in the upper part

80 mm edge, 266 mm height

Folie 25







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Comparison of different ADDmeth designs

• Comparison of catalyst volume (leads to higher synthesis power)







Comparison of different ADDmeth designs

• **Comparison of the cross-section** in the lower section of the reaction channel (enables higher temperature)





vbb. 18: Vergleich der ADDmeth Designs in der Schnittansich



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New ADDmeth3 designs

- Both reactors build upon the learnings from the ADDmeth1 experiments
- The catalyst bed per unit cell is significantly larger than in ADDmeth1, and the local residence time in the lower section of the reactor is significantly improved.



• The heat pipes are better tailored to accommodate the released reaction heat

Additional features:

- Improved heat transfer
- 3D-printed steam generator

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Result: Methanation reactor with in-situ co-reforming of tars



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Outlook

• After characterization in the laboratory, the presented 5 kW unit cell is scaled up by connecting multiple triangular cells to form a 20 kW cluster.









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